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TECHNICAL PROGRESS REVIEW

ISOTOPES AND RADIATION TECHNOLOGY

VOLUME 1, NO. 1

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DRAFT
TECHNICAL PROGRESS REVIEW
ISOTOPES AND RADIATION TECHNOLOGY
VOLUME 1, NO. 1

Editors:
P. S. Baker
A. F. Rupp
R. H. Lafferty
M. W. Gerrard

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by
Isotopes Development Center

OAK RIDGE NATIONAL LABORATORY
Oak Ridge, Tennessee
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A new Quarterly Technical Progress Review, "Isotopes and Radiation Technology," was approved by the U. S. Atomic Energy Commission in the summer of 1962 and its preparation was assigned to ORNL as a part of the general program of the Isotopes Development Center. The assembly of a staff began almost immediately. By early fall, under the auspices of the AEC's Divisions of Technical Information and Isotopes Development, the preliminary drafts which are now the first issue of this periodical were being developed. "Isotopes and Radiation Technology" is the fifth of a series of Commission-sponsored quarterly technical progress reviews which cover broadly the field of nuclear science and technology. The others are Nuclear Safety, Power Reactor Technology, Reactor Materials, and Reactor Fuel Processing.

This new review will emphasize the reporting of DID-sponsored research and development. By publicizing present applications, it can help to expand the utilization of isotopes; and by helping the Commission and contractors to keep abreast of related activities within the AEC complex it can help prevent overlap of research effort and can aid in predicting isotope usage so as to point the direction for future isotope production efforts.

"Isotopes and Radiation Technology" is a technical progress review. Even guest articles will be of a review nature.* The editors and the AEC have no intention of duplicating the present efforts of commercial publishers of abstract journals and periodicals covering original research, and overlapping will be avoided. Books, monographs, and bibliographies that are considered pertinent to the areas routinely covered will be noted and occasionally reviewed.

*e.g., see report by G. H. Cartledge on technetium in this issue, p. 3.

TECHNICAL PROGRESS REVIEW
Isotopes and Radiation Technology

| | |
|--|------|
| Editorial | iii |
| Feature Article: | |
| INHIBITION OF CORROSION BY THE PERTECHNETATE ION | 1 |
| References | 12 |
| I. PRINCIPLES OF ISOTOPE TECHNOLOGY | 16 |
| Introduction | 17 |
| Isotope Production | 19 |
| Isotope Utilization | 20 |
| History and Chronology | 22 |
| References | 28 |
| STATUS OF THE UNITED STATES PROGRAM ON ISOTOPES AND RADIATION DEVELOPMENT | 39 |
| II. ISOTOPE PRODUCTION AND DEVELOPMENT | 115 |
| Fission Products | 116 |
| Stable Isotopes | 140a |
| References | 174a |
| III. APPLICATIONS TECHNOLOGY | |
| Analysis | 176 |
| Radiometric | 177 |
| References | 206 |
| Activation Analysis | 209 |
| References | 240 |
| Radioisotope Sources for Radiography | 245 |
| References | 253 |

| | |
|--------------------------------------|-----|
| References | 366 |
| IX. AVAILABLE MATERIALS AND SERVICES | 368 |
| X. BOOK REVIEWS | 372 |

VII. APPLICATIONS IN MEDICINE

The literature on medical uses of radioisotopes is briefly reviewed here.* Since developments in this rapidly expanding field have been summarized in various well-documented publications,¹ earlier original literature was not extensively consulted. Only a few of the many published papers are listed here, and emphasis is on recent, rather than older, work. Since many reports are preliminary and opinions on the effectiveness of some of the isotopes differ widely, in future issues of this quarterly various viewpoints on the medical use of radioisotopes will be presented, and some of the less well-known isotopes, both old and new, will be reviewed. Readers with publications in this field are invited to send in reprints to ensure coverage.

The medical use of artificial radioactive isotopes began in the late thirties.² The increasing interest in them is indicated by the numerous conferences,³ journals,⁴ bibliographies,⁵ and reviews⁶ partly or entirely devoted to this field. Radioisotopes are used in both diagnosis and therapy (Table VII-1⁷), in the former in extremely small amounts; in the latter, chiefly in malignant or malignancy-related conditions. Safety in their use is ensured (1) by requiring the licensing of physicians who use them -- and then only after special training; (2) by standardization of handling techniques and adoption of special safety regulations by hospitals.⁸ Early fears of long-term adverse

*The editors of this review gratefully acknowledge the many helpful suggestions made by W. G. Myers, M.D., and M. Brucer, M.D., who reviewed a rough draft of the manuscript. Some of their suggestions that could not be readily incorporated will be used in future issues.

Table VII-1. Radioisotopes Used in Diagnosis and Therapy*

| Diagnosis | | |
|---|---|--|
| Test | Measurement | Usefulness |
| I^{131} studies of thyroid function; NaI, usually oral | Percentage of dose concentrated in thyroid, percentage of dose excreted in urine; plasma free and bound I^{131} | Valuable index of thyroid function; drugs cause interference |
| I^{131} for diagnosis of thyroid carcinoma; oral | Scanning and external procedures | Relatively little value while normal thyroid gland present; highly useful in locating well differentiated metastatic lesions after thyroidectomy |
| Fe^{59} iron metabolism; usually intravenous iron citrate; oral doses used to study absorption | Serial plasma assays for several hours; red cell assays for about 10 days; external counts over spleen, liver, sacrum; blood and fecal assays after oral dose | Helps explain mechanism of some anemias, importance of impaired erythropoiesis versus hemolysis; oral test indicates absorption defects. |
| Cr^{51} red cell survival; red cells labeled with Cr^{51} in vitro and injected intravenously | Serial assays of blood (red cells) for 2 or 3 weeks; external counts over spleen, liver, heart; fecal assays for gastrointestinal bleeding | Indicates rate of hemolysis and to some extent the role of the spleen in hemolysis |
| Co^{60} -vitamin B ₁₂ ; oral | Fecal and urinary excretion; sometimes external counts over liver | Shows fundamental defect of pernicious anemia even if patient is not in hematologic relapse |
| I^{131} -labeled rose bengal; intravenous | Rates of removal from bloodstream and concentration in liver | An index of liver function and biliary patency |

Table VII-1. Contd.

| Test | Measurement | Usefulness |
|---|--|---|
| I^{131} -labeled fat; oral | Blood, urinary, and fecal levels | Information on intestinal absorption, pancreatic function |
| Cr^{51} , p^{32} red cell mass; I^{131} albumin-plasma volume, Na^{24} , K^{42} for Na and K spaces | All based on dilution principle; solution of isotope (or suspension of red cells tagged in vitro) given intravenously and after mixing in the body, blood concentration determined | These tests give data on red cell and plasma volumes, fluid balance and electrolyte status; may be helpful in presurgical evaluation, study of metabolic and endocrine disorders. |

| Therapy | | |
|--------------------------------|---|---|
| | Patients Treated | Usefulness |
| I^{131} as NaI; usually oral | Uncomplicated Graves' disease patients over 40; complicated Graves' disease any age | General results excellent; multiple doses needed in some cases; significant incidence of early or delayed myxedema |
| | Toxic nodular goiter in selected patients | Eventual control of toxicity in most cases; may require repeated doses; concern over malignancy already present at onset, plus desire for rapid result, may recommend surgery for patients who can be well prepared |

Table VII-1. Contd.

| Patients Treated | Usefulness |
|---|---|
| Thyroid carcinoma in selected patients | For treatment of relatively small group of patients with nonresectable lesions showing ability to concentrate the isotope |
| Certain patients with angina pectoris or intractable congestive heart failure | Clinical improvement in significant percentage |
| p ³² as soluble phosphate; oral or intravenous | Generally good results |
| Chronic granulocytic leukemia | Useful, often in conjunction with local irradiation |
| Chronic lymphocytic leukemia | Good results in patients not suitable for phlebotomy alone |
| Polycythemia vera | Occasionally some value |
| Lymphosarcoma, multiple myeloma, diffuse bone metastases | Rarely useful |
| Acute leukemia, Hodgkin's disease | Symptomatic value; partially supplanted by nitrogen mustards |
| Colloidal Au ¹⁹⁸ , p ³² , y ⁹⁰ | Very limited value compared with other forms of therapy |
| Intracavitary in patients with effusions and ascites caused by malignant tumors | |
| Intravenous for granulocytic leukemia and diffuse lymphoma of liver | |

Table VII-1. Contd.

| | Patients Treated | Usefulness |
|---|---|--|
| | Direct injection into tumor in certain cases of prostate carcinoma; attempt to radiate lymphatic channels | Believed useful if properly used in carefully selected cases |
| Co ⁶⁰ , Cs ¹³⁷ , radium-226, radon-222, Sr ⁹⁰ for implantation and brachytherapy | Certain localized tumors, especially carcinoma of cervix, head, and neck | Useful in skilled hands for properly selected patients |
| Co ⁶⁰ , Cs ¹³⁷ for teletherapy | For all lesions suitable for external radiation therapy | Highly useful and practical |

*Reprinted with minor style changes with the kind permission of publishers, from G. A. Andrews, Medical students and radioisotopes, The New Physician, 9(12):23-7 (1960).

effects on patients have been tempered by the results of experience. According to Lawrence,⁹ for example, cancer has not once been reported as a result of radioactive iodine therapy of thousands of hyperthyroid patients; and in 25 years, the survival rate of more than 500 polycythemia vera patients treated with P^{32} has been about the same as that of patients with diabetes or pernicious anemia treated by standard methods. There is wide usage of radioactive iodine for toxic goiter in patients over 45, of P^{32} for polycythemia vera, and of various isotopes for pituitary or thyroid ablations to relieve other conditions.¹

Among the radioisotopes now used in internal medicine in the largest amounts are I^{131} , P^{32} , Au^{198} , Cr^{51} , Fe^{59} , and Hg^{203} , but many others are used in smaller amounts; large quantities of Co^{60} and Cs^{137} are used in teletherapy units. Many radioisotopes are being studied for potential use, and useful ones are introduced into the field as new production methods are developed and costs are lowered. Criteria for suitable isotopes include high specific activity, suitable half-life, physiological compatibility with body processes, and the ability to at least partially localize in the tissue being examined or treated. Radioiodine, for example, while not entirely thyroid-specific, is an obvious choice for thyroid conditions, as is calcium for bone. Specificity may sometimes be obtained by incorporation in compounds (e.g., radiocobalt in vitamin B_{12} , which localizes in the liver). An isotope may be localized physically as a wire, as a colloid, or as the recently introduced ceramic beads.¹⁰ Both the physiological and biological half-life must be considered, and selection may be made on the basis of the need for particular radiation characteristics -- from a soft beta all

the way to a hard gamma radiation. In earlier work, isotopes with a short physical half-life were found inconvenient to use,¹¹ but they are finding application now with development of new instrumentation and sources of supply.

Iodine

I^{131} (8.1 d, 0.33- and 0.61-Mev β 's, 0.36- and 0.64-Mev γ 's).

Iodine-131 is most commonly used for diagnosis of thyroid disorders.¹ Other uses include incorporation in human serum albumin for determination of the total circulating plasma volume, placentography,¹² estimation of mitral valve size,¹³ and detection of right-to-left cardiac shunts. There has been partial success in localizing brain tumors with I^{131} -labeled compounds, and I^{131} -labeled methyl iodide has been used for detecting left-to-right cardiac shunts.¹⁴ Rose Bengal containing I^{131} is useful for liver function tests and for blood flow tests¹⁵ and has been suggested for detecting liver metastases that are not obvious from surgery.¹⁶ Recent reports indicate the Rose Bengal- I^{131} hepatoscan to be a satisfactory technique.¹⁷ Other I^{131} -labeled compounds reported useful are Hippuran for renal tests¹⁸ and for determining cerebral circulation time;¹⁹ Diodrast for evaluating venous diseases of the lower extremities;²⁰ sulfobromophthalein for liver function tests;²¹ and compounds for studying fat metabolism.²² An I^{131} -labeled compound demonstrated 200-ml pericardial effusions in a simulated system,²³ and an accurate and reproducible method for insulin assay with I^{131} has been reported.²⁴

In therapy, I^{131} has been used for nontoxic²⁵ as well as toxic goiter.¹ It is used as a palliative in thyroid carcinoma, and an

I^{131} -tagged fat-soluble medium, which accumulates in regional lymph nodes, is potentially useful for irradiation of neoplasms.²⁶ Thyroid ablation by I^{131} in more than 1000 patients improved angina pectoris in 75% and congestive heart failure in 60% of those afflicted.²⁷ Fortunately, gamma irradiation of the ovaries from urinary excretion of administered I^{131} is considered negligible.²⁸

I^{125} (57.4 d, no β , 0.035-Mev γ). Iodine-125^{*}, which is easily produced by neutron bombardment of Xe^{124} , is becoming increasingly valuable in diagnosis. Irradiation of a patient by it is about half²⁹ that from I^{131} , assuming that the biological half time is equal to the physical half-life. Good thyroid and liver scans have been made with it,³⁰ and Hippuran labeled with I^{125} has been used for renograms.³¹ As a tag for serum albumin and oleic acid and in the salt NaI^{125} , it has given good results in experimental blood volume, absorption, and thyroid tests,³² and has been suggested for indicating the extent of arteriosclerosis.³³ In a promising new technique for lipid digestion and absorption studies, a compound labeled with I^{131} and another labeled with I^{125} are administered simultaneously.³⁴ The promising use of I^{125} in radiography³⁵ is indicated by a photograph of the bones of a hand made with this isotope (Fig. R3,R4).. Exposure to both patient and technician is low because of the ease of shielding against the 27.2 and 35.4-keV photons. Portable x-ray equipment is being developed with I^{125} sources.³⁶

I^{132} (2.3 hr, 1.5- to 2.1-Mev β 's, 0.53- to 0.95-Mev γ 's).
The use of I^{132} , which is easily recovered from a tellurium source, would decrease patient exposure to 1-2% of that from I^{131} and would

* See also p. 245.

allow repeated administration.³⁷ Thyroid function after I^{131} therapy may be determined by observing the absorption of I^{132} , whose higher energy gammas are easily detected in the presence of residual I^{131} .³⁸

I^{123} (13.3 hr, no β , 0.16-Mev γ). With I^{123} , exposure of a patient would be only 1-5% of that from I^{131} . Experiments on preparation of curie amounts are projected.³⁹

I^{124} (4 d, 1.53- and 2.13-Mev β 's, 0.6- and 1.72-Mev γ 's).

Iodine-124 has been suggested for greater uniformity of dose in thyroid carcinoma.⁴⁰

Phosphorus

P^{32} (14.3 d, 1.71-Mev β , no γ). The largest use of P^{32} is in treatment of polycythemia vera and chronic leukemia.¹ It has not been very useful in acute leukemia. Blotting paper soaked with radioactive sodium phosphate was used in early applications for skin lesions, and activated red phosphorus incorporated in polyethylene in later work. Metastatic bone cancer has been treated with P^{32} as the phosphate.⁴¹

Colloidal chromic phosphate- P^{32} and Au^{198} , as well as nitrogen mustard, for control of malignant intracavity effusions are compared in a recent review⁴² of 222 cases treated with P^{32} and 2110 with Au^{198} . Colloidal chromic phosphate- P^{32} has been used for palliation of cancer, and colloidal zirconyl phosphate- P^{32} administered intravenously to more than 250 patients with chronic leukemia and lymphomas is reported to have resulted in prolonged remissions and long treatment-free periods.⁴³

In diagnostic applications, an in vivo test for human spermatogenesis using sodium phosphate- P^{32} has been described,⁴⁴ and the ability of newly proliferating cells to concentrate P^{32} has been used to

differentiate benign from malignant ocular lesions.⁴⁵

Gold

Au¹⁹⁸ (2.7 d, 0.97-Mev β , 0.41-Mev γ). Gold-198 is most commonly used as a colloid,¹ which has 0.003- to 0.007- μ -dia. particles. When this material is administered to a patient and a scan is performed, normal liver shows an even gradation of activity, but lesions result in activity concentrations.⁴⁶ Diagnostic uses include determination of liver blood flow,¹⁵ and in regions where the parasitic *Taenia Echinococcus granulosus* occurs it has been used for localizing hydatid liver cysts and for following up healing after drug administration or surgery.⁴⁷

Colloidal gold has been used in treating both primary carcinoma and metastases.⁴⁸ Transportation to the lymph nodes is hastened if the particles are coated with silver,⁴⁹ but radioactive colloidal gold has been injected directly into the lymphatics.⁵⁰ Other uses are in treating leukemia⁵¹ and in controlling malignancy-induced intracavity fluids.⁵²

Interstitial implants of Au¹⁹⁸ seeds have been used for hypophysectomy and for palliation of metastatic carcinoma.⁵³ The hypophysis implantation requires about 30 min. Problems in surgical and radiotherapeutic pituitary ablation are discussed in recent reviews.⁵⁴

Chromium

Cr⁵¹ (28 d, no β , 0.32-Mev γ). Hexapositive Cr⁵¹, in Na₂Cr⁵¹O₄, attaches to the red cells and may be used to determine such quantities as red cell survival time.¹ A new instrument for accurate blood volume measurement uses Cr⁵¹-tagged red cells or I¹³¹-albumin.⁵⁵ Such cells have been suggested for determination of cardiac output and for placental

localization (no activity is transferred to the fetal circulation), and are the basis for a simple method for quantitation of menstrual blood loss⁵⁶ and for spleen studies.⁵⁷ Tripositive chromium (as $\text{Cr}^{51}\text{Cl}_3$) in solution attaches to plasma proteins and is useful for determining plasma volume. Serum albumin labeled with Cr^{51} is easily prepared and has been introduced for demonstrating gastrointestinal protein loss.⁵⁸

Implants of Cr^{51} have been recommended for cancer therapy.⁵⁹

Iron

Fe^{59} (45 d, 0.26- and 0.46-Mev β , 1.1- and 1.3-Mev γ). Detailed methods for Fe^{59} tests are described in a recent brochure.⁶⁰ As the chloride, citrate, or sulfate, Fe^{59} has been used for iron absorption and utilization studies.⁶¹ It has also been used for determining red cell survival time, but the necessity of labeling the cells in vitro makes it less convenient than Cr^{51} . The advantages and disadvantages of Fe^{59} and Cr^{51} are discussed in a recent review.⁶²

Fe^{55} (2.94 y, 0.0059-Mev x ray following electron capture).

Comparison of the percentage utilization of intravenously administered Fe^{59} to that of oral Fe^{55} has been used for iron absorption studies.⁶³

Mercury

Hg^{203} (47 d, 0.21-Mev β , 0.28-Mev γ). Besides the use of mercury isotopes in tracing kidney drugs,⁶⁴ Hg^{203} has been used for localizing brain tumors and for determining cerebral circulation time.⁶⁵ Neohydrin and Chlormerodrin labeled with Hg^{203} have been used for kidney examinations.⁶⁶

Boron

B¹⁰. When the stable isotope B¹⁰ is deposited in brain tumor tissue and activated by neutron bombardment, the released alpha radiation is effective for control and palliation.⁶⁷ By this treatment, cells of transplantable sarcomas in mice have been killed and cancer has disappeared in 2-4 weeks.⁶⁸ This method of treatment is not new, but basic techniques are still being investigated.⁶⁹

Fluorine

F¹⁸ (1.87 hr, 0.65-Mev positron emission). Fluorine-18 has been used as a replacement for Ca⁴⁷ and Sr⁸⁵ in bone scanning.⁷⁰ It has also found application in thyroid studies⁷¹ and in locating brain tumors.⁷²

Selenium

Se⁷⁵ (120 d, no β , 0.14- to 0.28-Mev γ 's). The high specificity of Se⁷⁵-selenomethionine for the pancreas suggests that this compound might be useful in early detection of pancreas carcinoma.⁷³

Krypton

Kr⁸⁵ (10.6 y, 0.67-Mev β). Because of its low solubility, injected or inhaled Kr⁸⁵ is useful in determining cerebral and coronary blood flow and cardiac and pulmonary shunts.⁷⁴ For example, if a right-to-left cardiac shunt is present, activity in the arterial system rises rapidly when Kr⁸⁵ is given intravenously.⁷⁵ If a left-to-right shunt is present, the Kr⁸⁵ concentration in the pulmonary artery reaches a value 20 times that in a systemic artery following inhalation of this gas.⁷⁶ A simple technique for studying pulmonary function with Kr⁸⁵ has been devised,⁷⁷ and a new technique for rapid and accurate

measurements of serial heart blood flow was reported recently.⁷⁸

Strontium and Yttrium

Sr^{90} (28 y, 0.54-Mev β , 2.27-Mev β from Y^{90} daughter, no γ).

In an external source containing Sr^{90} designed for use in ophthalmology, the Sr^{90} beta particles are filtered out so that only the Y^{90} betas actually reach the lesion. The lens is not irradiated.¹

"Radiation curettage," destruction of the uterine mucosa by radiation from a Sr^{90} source placed in the uterine cavity, has been recommended for control of endometrial bleeding when no malignancy is indicated.⁷⁹

Sr^{85} (65 d, no β , 0.5-Mev γ). Strontium-85 has been used for localizing bone tumors⁸⁰ and for studying bone metabolism. Because of its shorter half-life, irradiation of the body is less than with Sr^{90} . Its gamma radiation can be detected externally.

$\text{Sr}^{87\text{m}}$ (2.8 hr, 0.39-Mev γ). This isotope, which can be obtained quickly, as needed, from 80-hr Y^{87} , has been mentioned⁸¹ for possible use in obtaining information available in a short time; for example, circulation time, cardiac output, and various clearances, and in radiography.

Y^{90} (64.5 hr, 2.27-Mev β). Injected ceramic Y^{90} particles of ~60 μ dia., which are trapped in capillaries of the lung, liver, and other organs, have been used for cancer therapy.⁸² Thirty-five patients with Parkinson's disease were improved by brain implants of ceramic Y^{90} beads.⁸³ Carrier-free Y^{90} may be eluted with citrate from a cation exchange resin preparation of Sr^{90} .⁸⁴ It may be infiltrated as a solution, implanted in a plastic filament, or applied in a bag or absorbed in

blotting paper.⁸⁵

Because it is a pure beta emitter, colloidal Y^{90} results in less exposure of hospital personnel than Au^{198} , and it has been used for control of malignancy-induced effusions.⁸⁶ Complete pituitary destruction has been reported as a result of packing the fossa, following surgery, with powdered Y^{90} dispersed in wax.⁸⁷ For superficial radiation therapy, Y^{90} has been incorporated in small plates, which can be selected to cover an irregular area.⁸⁸ Transperineal injection of Y^{90} has been used for prostatic cancer.⁸⁹ Intravenous Y^{90} gave good results in eight polycythemia vera patients.⁹⁰

Calcium

Ca^{45} (164 d, 0.25-Mev β , no γ). Calcium-45 has been used for studying calcium metabolism, Paget's disease of bone, and hypoparathyroidism.⁹¹

Ca^{47} (4.7 d, 0.66- and 1.94-Mev β 's, 0.48- to 1.3-Mev γ 's; accompanied by 3.5 d Sc^{47} daughter with 0.46- and 0.62-Mev β 's and 0.175-Mev γ). Possible medical uses of the now-available Ca^{47} , which is of interest because of its high gamma energy, are being extensively investigated.⁹²

Cobalt

Co^{60} (5.3 y, 0.31-Mev β , 1.17- and 1.33-Mev γ). By far the largest use of Co^{60} is in teletherapy units,^{1b,c,93} descriptions of which may be obtained from various equipment manufacturers (see also book review section of this issue). Such units were widely distributed by 1959 (Table VI-2).⁹⁴ In 1962 they were reported to number 1400, of which 550 are in the U.S.A. and Latin America.⁹⁵ They are reported⁹⁶

Footnotes to Table VI-2.

^aReprinted, with minor style changes, from reference 94.

^bThese figures represent the number of licenses issued as of August 1, 1959; they do not include such teletherapy installations as those at Oak Ridge Institute of Nuclear Studies and Argonne Cancer Research Hospital, for which no license was required.

^cEstimated, mainly low curie units.

^dIncluding only countries for which information is available.

Table VI-2. Radioisotope Teletherapy Source Distribution

| Location | Unit | | | Location | Unit | | |
|-----------------------|-------------------|-------------------|------------------|--------------------------|-------------------|-------------------|------------------|
| | Ir ¹⁹² | Cs ¹³⁷ | Co ⁶⁰ | | Ir ¹⁹² | Cs ¹³⁷ | Co ⁶⁰ |
| <u>Africa</u> | | | | Lebanon | - | - | 1 |
| Algeria | - | - | 1 | Philippines | - | - | 4 |
| Union of South Africa | - | - | 1 | Thailand | - | - | 1 |
| <u>America</u> | | | | <u>Europe</u> | | | |
| Argentina | - | - | 3 | Austria | - | - | 2 |
| Brazil | - | - | 8 | Belgium | - | - | 2 |
| Canada | - | 2 | 24 | Denmark | - | - | 2 |
| Chile | - | - | 2 | Finland | - | - | 1 |
| Cuba | - | - | 2 | France | - | 1 | 40 |
| El Salvador | - | - | 1 | Germany (Fed. Republic) | - | - | 4 |
| Mexico | - | - | 12 | Greece | - | - | 3 |
| Peru | - | - | 1 | Hungary | - | - | 1 |
| Puerto Rico | - | - | 2 | Italy | - | - | 26 |
| United States | - | 6 ^b | 264 ^b | Monaco | - | - | 1 |
| Uruguay | - | - | 2 | Netherlands | - | - | 4 |
| Venezuela | - | - | 5 | Norway | - | 1 | - |
| | | | | Portugal | - | - | 1 |
| <u>Asia</u> | | | | Spain | - | - | 5 |
| Burma | - | - | 1 | Sweden | - | - | 7 |
| Ceylon | - | - | 1 | Switzerland | - | - | 4 |
| China (Taiwan) | - | - | 1 | United Kingdom | 1 | 8 | 33 |
| Hong Kong | - | - | 1 | Yugoslavia | - | - | 1 |
| India | - | - | 3 | | | | |
| Indonesia | - | - | 1 | <u>Oceania</u> | | | |
| Iran | - | - | 1 | Australia | - | - | 4 |
| Israel | - | - | 3 | New Zealand | - | - | 2 |
| Japan | - | - | 200 ^c | <u>Total^d</u> | 1 | 18 | 689 |

See following page for footnotes.

to be in use in Communist China, and have been referred to as the most important product of Atomic Energy of Canada Ltd.⁹⁷ A nonroutine use of Co^{60} teletherapy is a possible new treatment for leukemia in which blood, circulated outside the body through a tube, is exposed to Co^{60} radiation.⁹⁸

Cobalt-60 needles have been implanted in tumors, and seeds enclosed in nylon tubing have been used in interstitial therapy and in wounds after surgery for a malignant condition.⁹⁹ Spheres of Co^{60} , gold-plated to prevent dusting, may be placed in hollow organs such as the uterus or bladder.¹⁰⁰ Cobalt-60 may be incorporated in plastic and molded to fit irregular areas such as the oral cavity.¹⁰¹

Vitamin B_{12} labeled with Co^{60} has been used in diagnosing pernicious anemia¹⁰² and localizing liver tumors.¹⁰³

Other Cobalt Isotopes. Co^{56} (77.3 d, 0.44- and 1.46-Mev β , 0.85-1.75-Mev γ); Co^{57} (270 d, no β , 0.014- and 0.12-Mev γ); Co^{58} (71 d, 0.48-Mev β , 0.81-Mev γ). Various other cobalt isotopes have been used¹ to label vitamin B_{12} . Cobalt-57 is reported¹⁰⁴ to give the lowest irradiation of a patient. Labeled vitamin B_{12} (Co^{57}) has been shown to concentrate in the parathyroid in dogs¹⁰⁵ and is therefore potentially useful in parathyroid conditions.

Cesium

Cs^{137} (30 y, 0.51- and 1.18-Mev β , 0.662-Mev γ from daughter Ba^{137}).

This isotope, which is recovered as a by-product from the processing of nuclear reactor fuel, is a relatively new isotope, but many Cs^{137} teletherapy units are in use (Table VII-2). The long half-life and low cost of the activity are the principal advantages over Co^{60} . Typical

units have been described elsewhere.^{1c,93a,106} (See also book review section of this issue.)

Iridium

Ir¹⁹² (74 d, 0.67-Mev β , 0.30- to 0.47-Mev γ 's). Radioiridium teletherapy is an old idea.¹⁰⁷ A new 100-curie Ir¹⁹² irradiation unit weighs only 38 lb; a 10-curie source weighing 28 lb is also available.¹⁰⁸ A unit with interchangeable ends of different sizes and shapes, for interorificial therapy, has been described recently.¹⁰⁹

The half-life of Ir¹⁹² is about 5% that of Co⁶⁰, and its less penetrating gamma rays allow greater local absorption and necessitate less protection for hospital personnel. Internally, it has been used as seeds embedded in nylon ribbon or as needles.¹¹⁰ Experimentally, an Ir¹⁹²-labeled resin suspension has been used with external counting for determining gastric emptying rates.¹¹¹

Arsenic and Copper

As⁷⁴ and Cu⁶⁴ (17.4 d and 12.8 hr, respectively; positron emitters).

These isotopes have been used for localizing brain tumors.¹¹²

As⁷⁶ (26.5 hr) and As⁷² (26 hr). Arsenic-76 has been used for brain tumor localization, as has As⁷², which yields nearly three times as many positrons per average disintegration¹¹³ as As⁷⁴.

Gallium

Ga⁷² (14.3 hr, 0.64- and 0.96-Mev β 's, 0.63- to 2.20-Mev γ 's).

Gallium-72 has been used to localize malignant bone tumors by external measurements, but the short half-life results in a concentration of this chemically toxic material which may perhaps be higher than is warranted by the radiation delivered.¹¹⁴

Ga⁶⁸ (68 min, 1.9-Mev β , 1.1-Mev γ). Gallium-68 has been suggested for such applications as brain tumor localization by annihiscopy. A source from which Ga⁶⁸ can be easily and rapidly obtained has been designed.¹¹⁵

Palladium

Pd¹⁰⁹ (13.5 hr, 1.03-Mev β) and Pd¹⁰³ (17 d). A mixture of these two isotopes has been used for interstitial irradiation.¹¹⁶

Tantalum

Ta¹⁸² (115 d, 0.36- to 0.51-Mev β 's, 1.12- and 1.22 Mev γ 's). Tantalum-182 wires have been used as tissue implants in control of malignant conditions.¹¹⁷

Ruthenium-Rhodium

Intragastric beta irradiation with Ru¹⁰⁶-Rh¹⁰⁶ has been tried.¹¹⁸

Lutetium

Lu¹⁷⁷ (175 d; also 6.8 d, 0.50-Mev β , 0.18- and 0.38-Mev γ). Lutetium-177 has been suggested for infiltration to lymph nodes involved in malignant disease.¹¹⁹

Sulfur

S³⁵ (87 d, 0.167-Mev β , no γ). Sulfur-35 has been used for determining extracellular water and as a therapeutic agent in chondrosarcoma and mycosis fungoides.¹²⁰ Labeling of Congo red dye with S³⁵ decreases the amount of dye needed for diagnosis of amyloidosis, eliminating the side effects that occur in some patients.¹²¹ Methionine labeled with S³⁵ has been suggested for use in detecting pancreas tumors.¹²²

Sodium

Na²² (2.6 y, 0.54-Mev β , 1.3-Mev γ) and Na²⁴ (15 hr, 1.4-Mev β ,

2.7- and 1.4-Mev γ 's). These isotopes have been used for measuring extracellular water, sodium space, and exchangeable sodium; for determining when skin grafts are complete; and for cancer therapy.^{1,123} Because of its shorter half-life, Na^{24} has advantages for some uses.

Potassium

K^{42} (12.4 hr, 3.6-Mev β , 1.5-Mev γ) and K^{43} (22 hr, 0.3- and 0.27-Mev β 's, 0.37- to 0.62-Mev γ 's). These isotopes have been used to localize brain and breast tumors, measure exchangeable potassium, and study potassium uptake by red blood cells.^{1,124} Less shielding is required with K^{43} because of its lower energy gamma radiation.

Rubidium

Rb^{86} (19 d, 1.78- and 0.72-Mev β 's, 1.1-Mev γ). Since rubidium acts similarly to potassium and has a more convenient half-life than K^{42} , it has been suggested for use in determining exchangeable potassium.^{124c} However, it must be remembered that Rb^{86} is not a potassium isotope. Myocardial scanning tests have been made with Rb^{86} in dogs.¹²⁵

Bromine and Chlorine

Br^{82} (35.7 hr, 0.44-Mev β , 0.55- to 0.78-Mev γ 's). Bromine-82 ratios in blood and spinal fluid have been used to differentiate between tuberculous and nontuberculous meningitis.¹²⁶ For determining

Ag^{111} was one of the early radioactive colloids used.¹²⁸ Silver-111, injected into the bloodstream as a protein-silver complex, is removed less rapidly by a cirrhotic liver than by a normal one.¹²⁹

Zinc

Zn^{63} (38 min, 1.38- to 2.34-Mev β 's, 0.67- and 0.96-Mev γ 's).

Zinc-63 was used in early work for control of malignant effusions.¹

Its use in studying zinc metabolism was reported recently.¹³⁰

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VIII. SAFETY